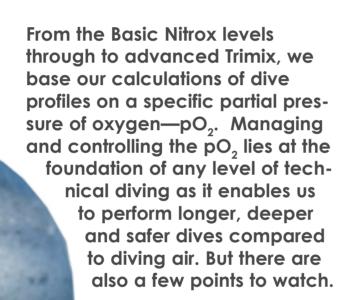
Cunningham

I survived the earlier stages of my diving career due to one reason - "luck".

# Manage your po



We are all aware that as we descend the gas in our lungs and body tissues compress and becomes denser. The deeper we go and longer we stay, the more of the inert gas Nitrogen (and in the case of diving with trimix, also helium) our tissues will absorb and hold. Oxvaen on the other hand is utilized in the metabolism, with carbon dioxide (CO<sub>a</sub>) being a resulting waste product. So, first of all, while diving deep and breathing gasses with elevated pO<sub>a</sub>, we want to keep our levels of exertion down as much as possible and metabolism and CO<sub>a</sub> production down. As work load increases, levels of CO<sub>2</sub> could

reach dangerous levels as it leaves divers more prone to oxygen toxicity and nitrogen narcosis. (As well as giving you a very undesirable headache.)

## Weighing pros and cons

Breathing any gas at elevated partial pressures comes with both advantages and risks. The most obvious advantage is when you increase the pO<sub>a</sub> and conversely lower your pN<sub>2</sub> (partial pressure of Nitrogen) you reduce nitrogen absorption, and with it, all the issues around decompression. As every Basic Nitrox diver will know, even modest increases in oxygen content can lead to significant longer no-decompression limits.

On the other hand, each individual gas may cause, in some cases similar, in other cases very different, potentially dangerous symptoms that could lead to the diver failing to return to the surface.

# Know the partial pressures

That is, if we don't know how to manage the partial pressures our breathing gases. Using Dalton's law, we can exactly calculate the partial pressures of oxygen, nitrogen and helium at any depth. We can thus also calculate our dosage of these gases.

And nitrogen isn't the only gas that can cause a problem when you breathe it for longer times at elevated pressure. Oxygen toxicity is, for one, another issue we need to get a grip on when venturing into any sort of technical diving.

Breathing oxygen at elevated partial pressures for an extended time can be toxic and lead to confusion and loss of muscle control, and ultimately seizures and convulsions, which under water will most likely result in drowning.

### Limits

The NOAA time limit for an single exposure to a pO<sub>a</sub> of 1.6 bar is 45 min. For a pO<sub>o</sub> of 1.3 bar the limit is at a somewhat more comfortable 180 mins. NOAA is the National Oceanographic Atmospheric Administration in the US. Based on their studies, all training agencies state the absolute maximum PO<sub>2</sub> breathed during a dive should not exceed 1.6 at any stage, with 1.4 being the recommended max PO<sub>2</sub> for the bottom phase.

# Personal limits

The exact PO<sub>2</sub> or point where O<sub>2</sub> has a toxic affect on the central nervous system will vary greatly from individual to individual, as well as an individual from day to day, or from dive to dive.

If the type of environment, equipment, mission or task changes from dive to dive, exertion levels could also change, hence, altering and perhaps increasing the amount of O<sub>a</sub> metabolized. It is unfortunately not possible to calculate and establish the exact amount of O<sub>a</sub> metabolized considering all these variables. So, if you think you may be exerting more than normal on a

specific dive, chose to breathe a lower pO<sub>a</sub> if you have the option. I would recommend lower than 1.3 bar during dives with a heavier work load.

NOAA have also established the fact that a time limit at specific pO<sub>2</sub>'s is another factor that must be taken into consideration; The higher the pO<sub>a</sub> the shorter the time limit would be before O<sub>2</sub> has a toxic effect.

# Oxygen is our friend

So, let's try to put things into perspective and recap. Oxygen is good, it's our friend. The higher the oxygen content in a nitrox mix, the smaller the proportion is of nitrogen.

Oxygen also helps to create a pressure gradient—or pressure difference, if you like-between the partial pressures of gases within the lunas in relationship to the partial pressure of gases absorbed



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within the tissues. It is this pressure difference that forces the inert gases to move out of the tissue – we term this inert gas elimination. The inert gasses are N<sub>a</sub> and He

and culprits behind decompression related problems such as the bends (aka DSC or DCI). The less inert gas in the mix, the less inert gas gets absorbed in the tissue, giving us the option to extend bottom time or to just take advantage of the extra safety less absorption of inert gas is giving us.

#### But...

If limits—pressure or time, or a combination of both—are exceeded, oxygen will be less forgiving than nitrogen or helium. Nitrogen Narcosis and High Pressure Nervous Syndrome (HPNS) the effect of Helium on the body under high pressure) do not lead to drowning or embolism as a bad hit of oxygen toxicity could.

However, as long as the oxygen dose (pO<sub>2</sub>) and time limits are not exceeded and we have good general diving technique (low levels of exertion), it's highly unlikely we would experience

> oxygen toxicity-related problems.

> > Having said that, NOAA did once succeed—with the help of the US Navv and in a dry chamber—to induce an oxygen seizure in a human subject at a pO<sub>o</sub> of 1.3. This is the lowest inspired pO<sub>2</sub> on record

that a oxygen seizure has been induced. Having said that, the dude in the chamber was overexerting to an extreme degree during the experiment. But even so, the experiment clearly emphasized the necessity for low exertion levels while breathing elevated pO<sub>2</sub>'s during a dive.

## And don't forget...

So it seems oxygen is the primary gas we need to keep within recommended limits.

But the wise diver should not forget and overlook the potential hazards related to nitrogen narcosis while focusing on issues with oxygen. In my experience in deep diving, I've seen many more divers having serious problems with nitrogen narcosis rather than oxygen toxicity. Recreational training agencies have put a depth limit on air diving to max 40m. This was established due to the fact that most divers (without specific deep water technical training) begin to act irrational beyond this depth, putting themselves and possibly other divers at risk, combined with the fact that the NDL becomes impractically short.

# Don't push it

Traditionally, the mix that divers sometimes push depth limits with is air! Few divers with any sort of technical training would consider pushing beyond the recommended depth with an enriched air/nitrox mixture due to the increased risk of oxygen toxicity. However, I've met many a recreational diver happy to push past 40 meters without specific deep diver training with air—in some cases, well beyond 60 meters. Although the pO<sub>2</sub> of air does not exceeded NOAA recommendations until we reach 66 meters. divers should take the effects of nitrogen narcosis seriously, or risk ending up as a statistic.

#### **HPNS**

much more complicated by adding a third gas, helium, which has it's own list of potentially dangerous symptoms when the helium content is high and the descent was very fast. It is not something that the mix diver needs to consider in the first 100m. But for the more experienced mix diver pushing past 100m, the effects of **HPNS (High Pressure Nervous** Syndrome) could become an issue. Symptoms include vomiting, diarrhea, micro sleep and probably the most common "tremors", normally starting in the arms below the elbow. A slower rate of compression (descent rate) will reduce the severity of the tremor, but obviously a too slow descent will increase decompression time and overall gas needed.

For the tec diver using trimix,

the gas toxicity story becomes

Some deep mix divers choose to increase the nitrogen content below 100m, as nitrogen tend to act as a buffer, reducing the severity of the tremor. A sensible balance in this case is the key. Shaking around at depth is definitely something we would like to avoid but not to the point where clear logical thought processes and problem solving gets impaired due to too high nitrogen dose!

**Leigh Cunningham** is the technical manager and TDI Instructor Trainer for Ocean College, Sharm El Sheikh. Probably best known for his records - Leigh once held the record for the deepest dive in the Red Sea, and is the current holder of the record for deepest wreck dive - and attempts of reaching extreme depths, he also has a wide range of teaching credentials to his curriculum: TDI instructor trainer, DSAT Tech Trimix instructor, PADI MSDT IANTD Technical diver instructor

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